

# **Overview of Proven and Emerging CO<sub>2</sub> Purification Processes for CCUS**

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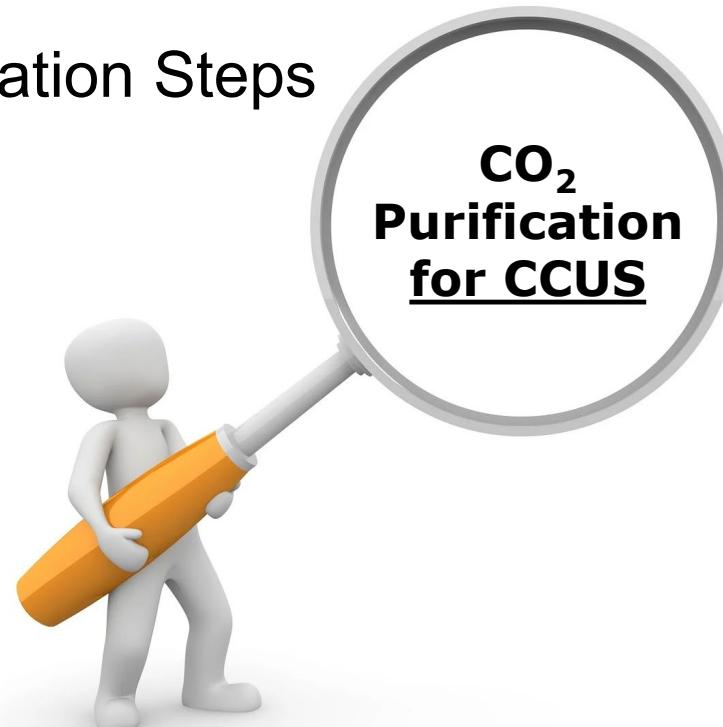
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Trimeric Corporation**

October 23, 2024

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# Presentation Roadmap

- Sources and Contaminants
- Sequence (Timing) of Purification Steps
- Dehydration
- Distillation
- Catalysts
- $\text{SO}_2$  / Other Sulfur Species
- $\text{NO}_x$
- Others: Aerosols, Salts, Mercury...



# High Purity Sources: Ethanol, Fertilizer, Amine Regen.

- $\geq 95\text{ %v. CO}_2$  (dry) basis
- Saturated with Water Vapor
- $\sim$ Atmospheric Pressure

→ **Compress and Dehydrate**



- High Oxygen Spikes ( $\sim 10,000\text{ ppmv}$ ) Common in Ethanol Plant CO<sub>2</sub>

→ **Distillation**



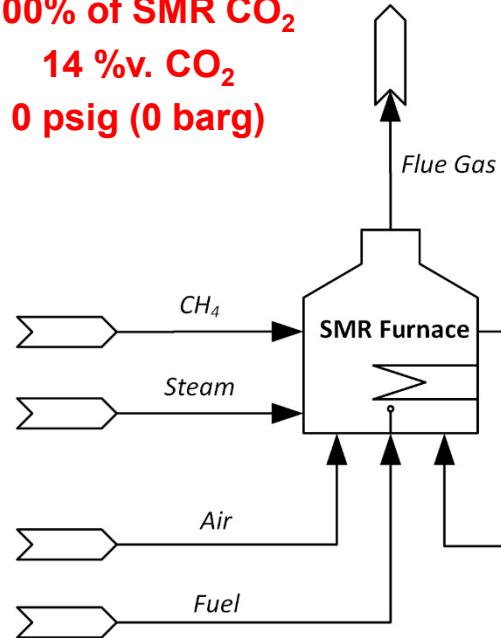
# Intermediate Purity Source: Hydrogen Production

## SMR Furnace Flue Gas

100% of SMR CO<sub>2</sub>

14 %v. CO<sub>2</sub>

0 psig (0 barg)



## PSA Feed (Shifted Syngas)

55% of SMR CO<sub>2</sub>

18 %v. CO<sub>2</sub>

350 psig (24 barg)

CO-Shift  
Reactor

PSA Feed

PSA Tail Gas

H<sub>2</sub>

PSA  
(H<sub>2</sub> Purification)

## PSA Tail Gas

55% of SMR CO<sub>2</sub>

40 %v. CO<sub>2</sub>

5 psig (0.4 barg)

SMR = Steam Methane Reformer



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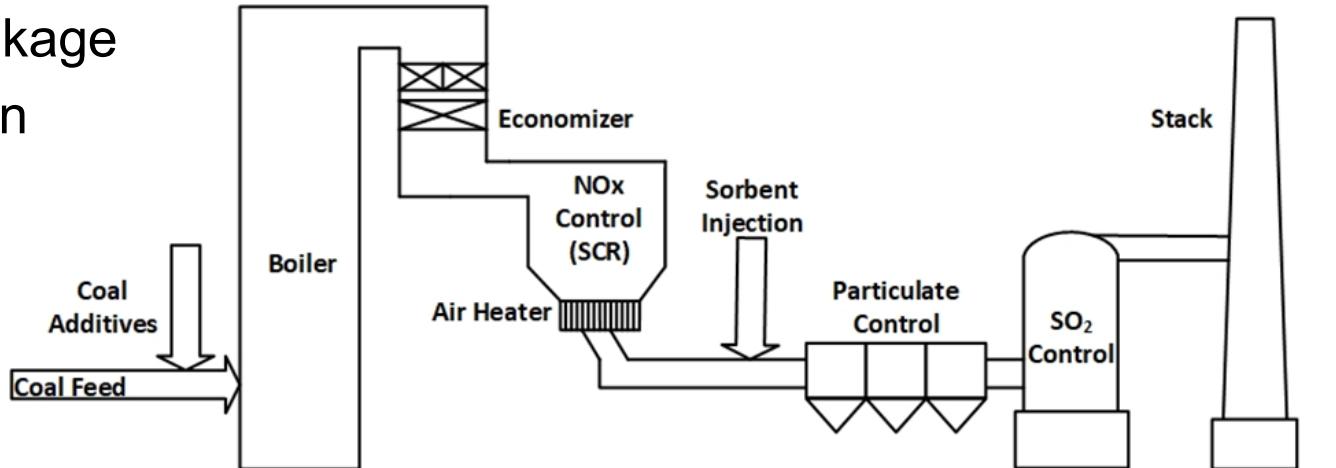
# Lower Purity Sources: Flue Gas

- ~Atmospheric Pressure, Water Saturated
- Too Hot for Capture Technologies
- **Contaminant Issues:**  
SO<sub>2</sub>, H<sub>2</sub>S, NO<sub>x</sub>, PM, Aerosols, Hg

Fuel / Process	Flue Gas ~%v.CO <sub>2</sub>
Natural Gas	4
Coal	12
FCC	15
Cement	21
Steel	26

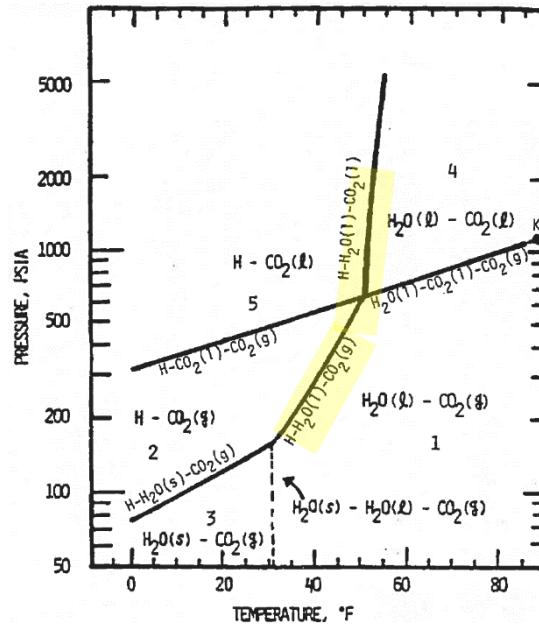
## Countermeasures Before Capture:

- Reduce Air In-Leakage
- Maximize Emission Controls →
- Quench and SO<sub>2</sub> Polishing



# Reasons for Dehydration of CO<sub>2</sub>

- Prevent Water Ice in CO<sub>2</sub> processed below 32°F (0°C).
- Prevent CO<sub>2</sub>–H<sub>2</sub>O Solid Hydrates → (Clathrates) above 32°F (0°C).
- CO<sub>2</sub> Corrodes Carbon Steel When Liquid Water Phase is Present.



**CO<sub>2</sub>-H<sub>2</sub>O Hydrate Figure**

Courtesy of Gas Processors Association.  
GPA Research Report RR-99

Corrosion of Carbon Steel Tube Sheet in Wet CO<sub>2</sub>, Water Condensing Service at 400 psig (28 barg).

# CO<sub>2</sub> Dehydration Process: TEG Absorption

- Typical performance ~ 150 ppmv H<sub>2</sub>O.
- Water Absorbed from CO<sub>2</sub> into TEG in Contactor.
- Heating TEG to 370°F (188°C) in Regeneration Skid Strips Water out of TEG.
- Suitable for Many Projects with only Compression Needed After Capture

## Concerns:

- Glycol Carryover in Dry CO<sub>2</sub>
- VOC/HAP Emissions in Vent Streams



Contactor



Regeneration Skid



# CO<sub>2</sub> Dehydration Process: Mole Sieve Adsorption

- Typical performance ~1 ppmv H<sub>2</sub>O.
- Process Gas Flows Top Down in Drying Bed Cycles.
- Heated Gas (400°F [204°C]) Flows up in Regeneration Bed Cycles.
- Capital and Operating costs ~1.5 times TEG Absorption.

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## Concerns:

- Media Must be Kept Free of Any Liquids (Water, Oil, etc.).
- Media Replaced ~Every 3 Years.

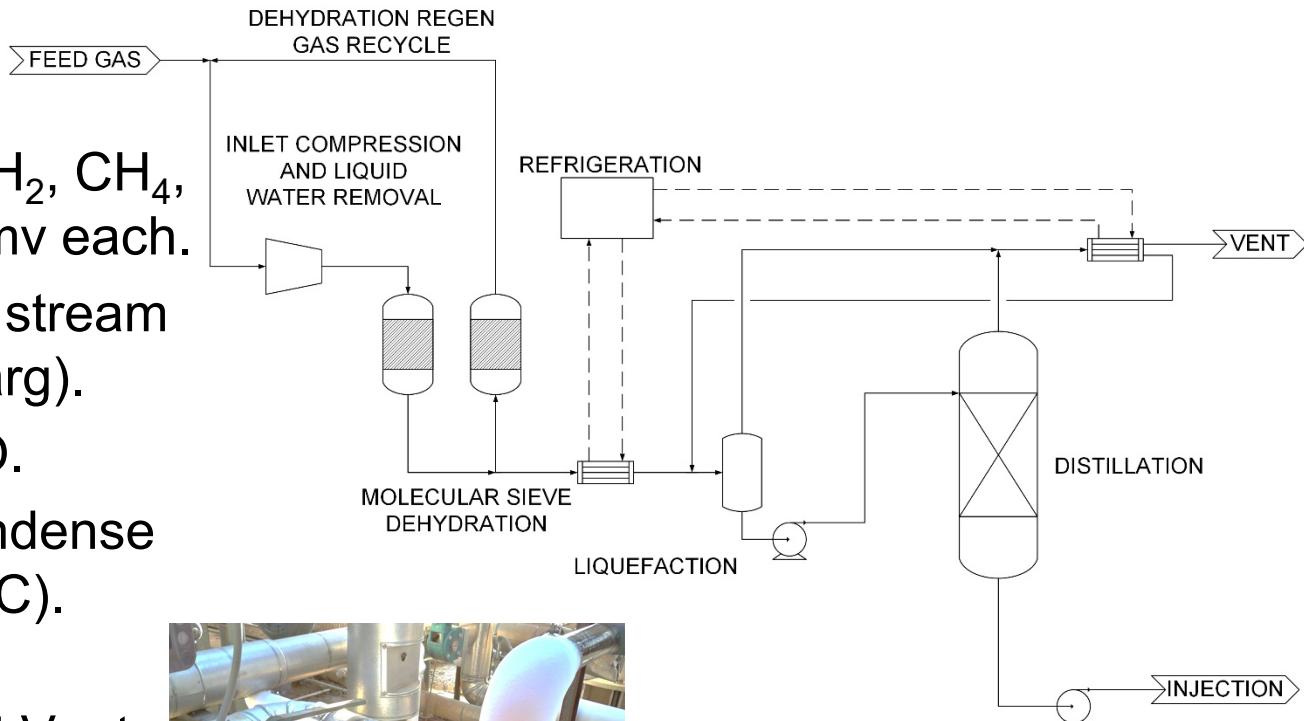


# Impacts of Oxygen in CO<sub>2</sub>

- Oxygen makes mixtures of CO<sub>2</sub> and water more corrosive.  
*(Remember the picture of the heat exchanger tube sheet on slide 6 ?!)*
- Oxygen can oxidize TEG leading to faster solvent degradation.
- Oxygen can lead to biological growth in underground formations.
- Oxygen can react with H<sub>2</sub>S under certain conditions to form elemental sulfur, sulfuric acid, and / or other sulfur compounds.

# Liquefaction and Distillation of CO<sub>2</sub>

- Removes Lighter Impurities: O<sub>2</sub>, N<sub>2</sub>, H<sub>2</sub>, CH<sub>4</sub>, CO, NO to <10 ppmv each.
- Compress CO<sub>2</sub> rich stream to ~ 350 psig (24 barg).
- Dry to ~1 ppmv H<sub>2</sub>O.
- Refrigeration to Condense CO<sub>2</sub> at ~10°F (−12°C).
- Lower Boiling Point Compounds Go Out Vent.



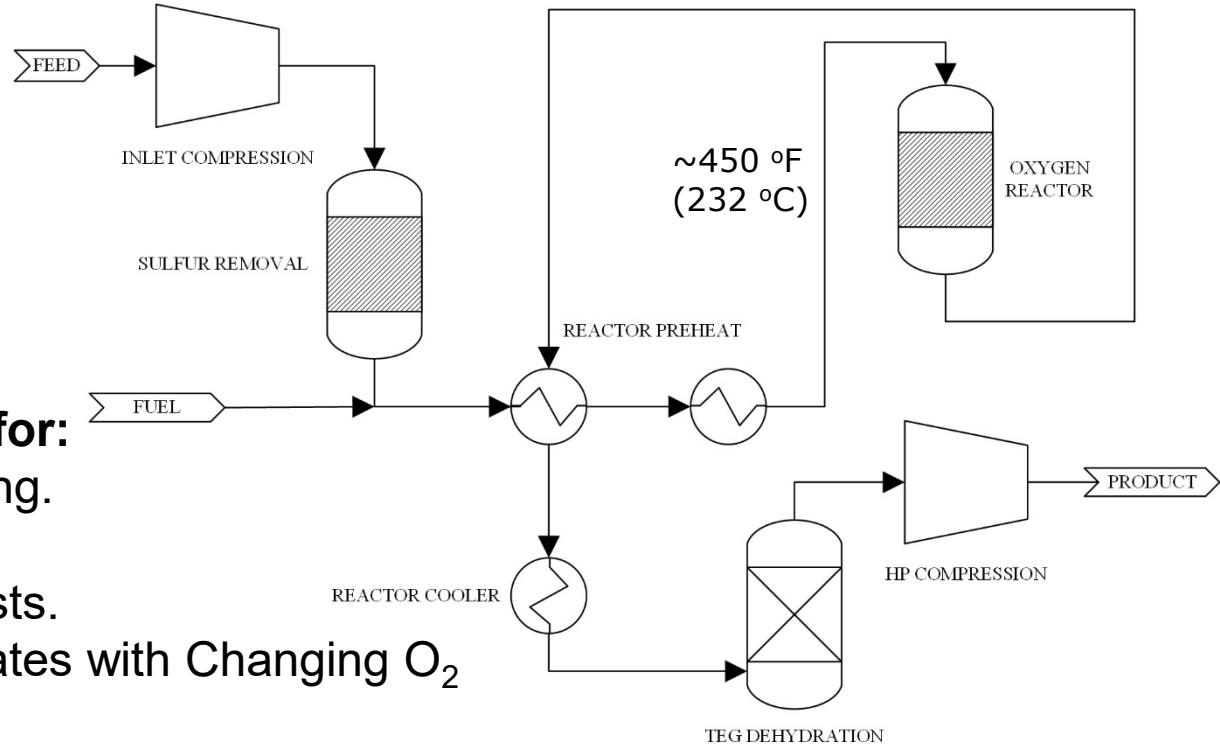
Water Vapor from the Air Freezes on Pipe Before Insulation is Installed.

# Emerging Option: Catalytic Reduction for Oxygen Removal from CO<sub>2</sub>

- Proposed for meeting 10 ppmv limit in CO<sub>2</sub>.
- Trimeric hasn't found any commercial applications.
- Demonstrated in other applications.

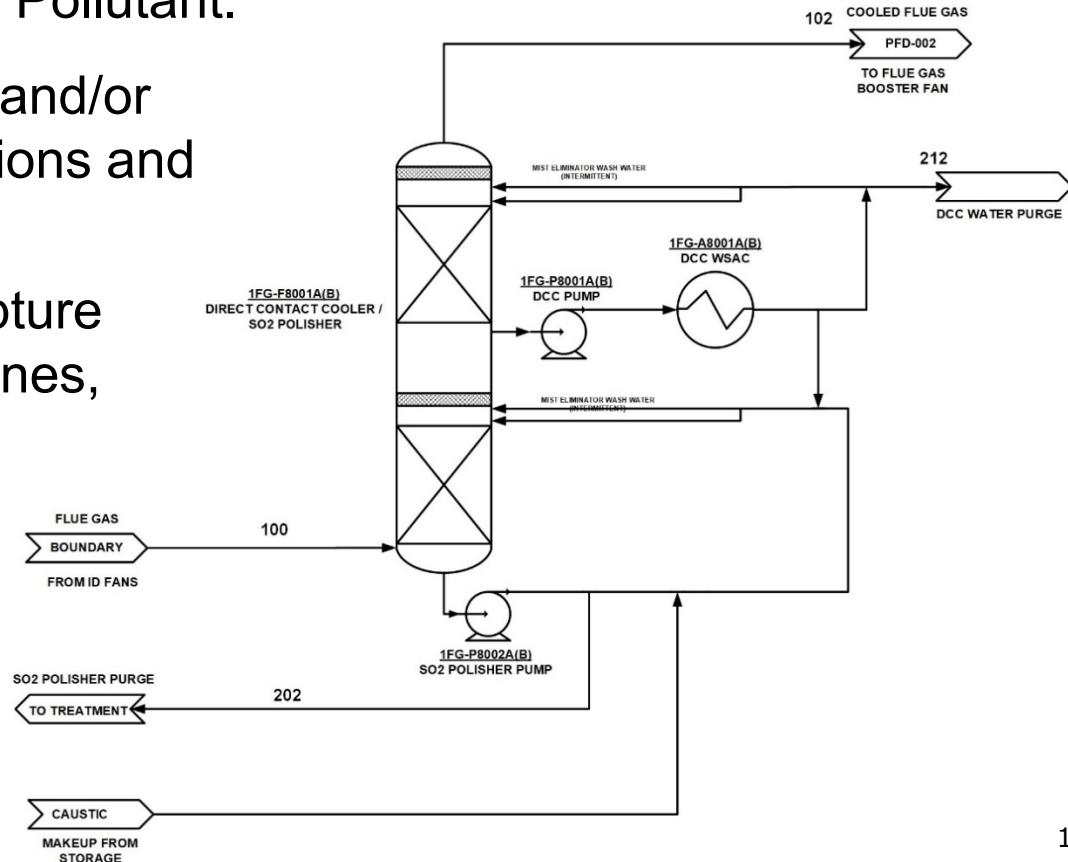
## Careful Evaluation Needed for:

- Catalyst Fouling and Coking.
- Catalyst Life and Sparing.
- Catalyst Replacement Costs.
- Matching Fuel Injection Rates with Changing O<sub>2</sub> concentrations in Feed.
- CO ppmv in Product vs. Spec Limits.



# Direct Contact Cooler / SO<sub>2</sub> Polisher (Caustic Scrubbing)

- SO<sub>2</sub> is Toxic and is a Criteria Pollutant.
- SO<sub>2</sub> can be Oxidized to SO<sub>3</sub> and/or H<sub>2</sub>SO<sub>4</sub> Causing Other Emissions and Corrosion.
- At the Front End of Most Capture Processes: Amines, Membranes, Sorbents
- Downstream of FGD unit, SO<sub>2</sub> Polisher can get down to 1 or 2 ppmv.



# H<sub>2</sub>S Removal Options for CO<sub>2</sub> Conditioning

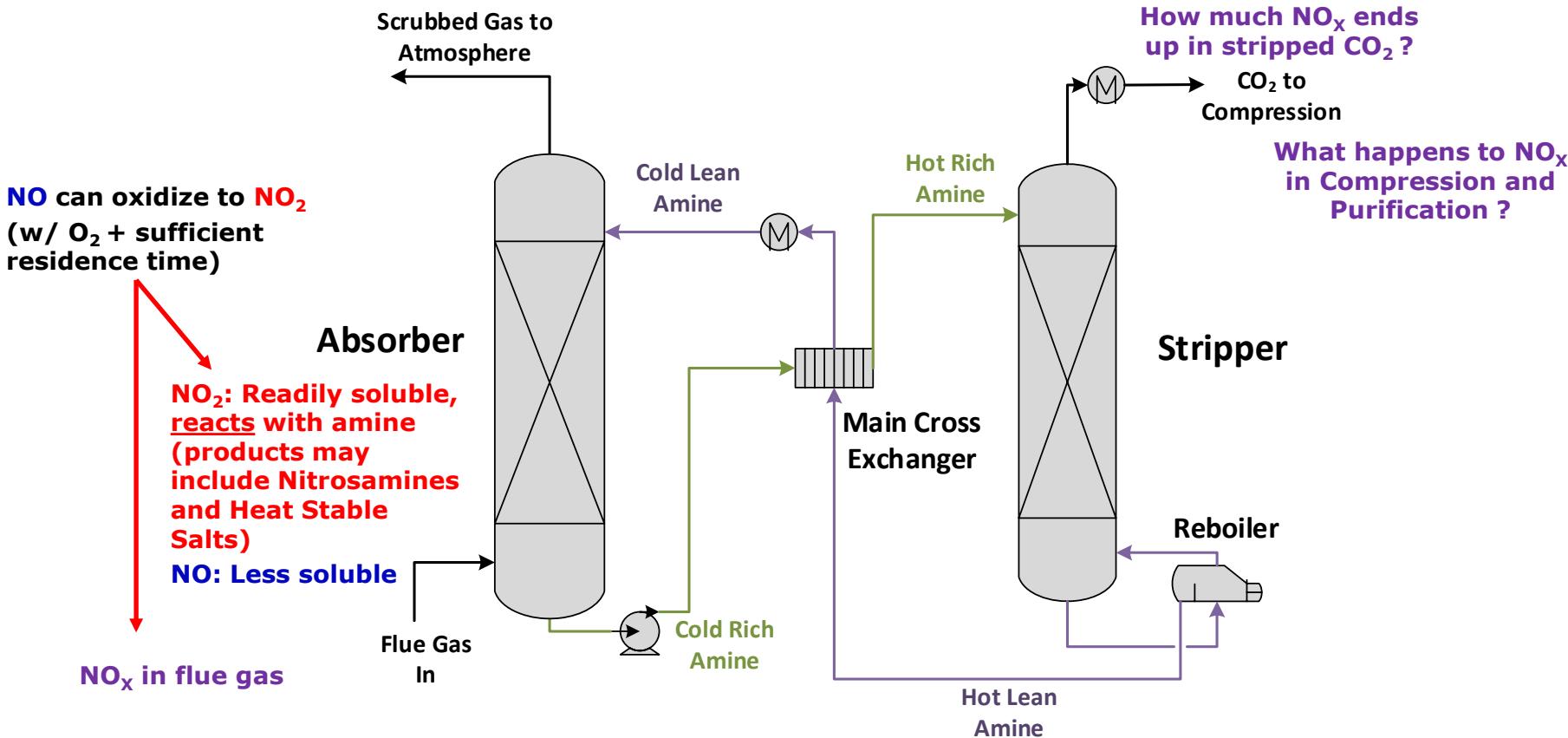
- ❑ H<sub>2</sub>S is Toxic and Corrosive.
- ❑ Technology Selection for removal from CO<sub>2</sub> depends on:
  - ❑ Feed Stream: Flow Rate, Pressure, Composition.
  - ❑ Treated Gas Specification.
  - ❑ Emission Limits for H<sub>2</sub>S and SO<sub>2</sub>.

Scale	Sulfur Loading (approx.)		Common Treatment Options
	lb/day	tonne/day	
Small	300	0.15	Scavengers
Medium	13,000	6	Liquid Redox
Large	45,000	20	Selective Amine, Claus

# NO<sub>x</sub> – General Background

- ❑ NO<sub>x</sub> = Nitric Oxide (NO) + Nitrogen Dioxide (NO<sub>2</sub>).
- ❑ NO<sub>x</sub> is a Criteria Pollutant, NO and NO<sub>2</sub> are Toxic and Corrosive.
- ❑ Present in Flue Gas Streams.
  - Coal-Fired Power Plant w/ SCR ~ 40 ppmv.
  - Gas-Fired Turbine Unabated ~ 100's of ppmv.
  - NO to NO<sub>2</sub> Ratio ~ 90 / 10.
- ❑ NO<sub>x</sub> may Lead to Specific Corrosion Risks in CO<sub>2</sub> Transport.
- ❑ NO<sub>x</sub> Must be Characterized in Detail for Each Source, Capture, and Purification Method.
  - (Example Next Slide).

# NO<sub>x</sub> in Flue Gas Example Case



# Flue Gas NO<sub>x</sub> – Manage Before Capture

Technology	Minimum Treated NO <sub>x</sub> Level	Challenges	Maturity
Water/Steam Injection	25 ppmv	Impact on turbine performance, high purity water needed	Commercially Proven
Dry Low NO <sub>x</sub> (DLN) Combustor Ultra Low NO <sub>x</sub> (ULN) Combustor	DLN: 9 - 15 ppmv ULN: 4 ppmv	Increased turbine cost/footprint, turndown challenges	Commercially Proven
Selective Catalytic Reduction (SCR)	~90% Reduction 5 ppmv	Ammonia slip, difficult to retrofit, catalyst management	Commercially Proven
NO <sub>x</sub> Scrubbing	Limited Use in This Application	Large liquid waste stream with nitrates/other chemicals, chemical costs	Developmental



# NO<sub>x</sub> – Manage During CO<sub>2</sub> Purification

Technology	Description	Key Challenges	Maturity
Distillation	NO Exits in Overhead Stream and NO <sub>2</sub> Exits in Bottoms Stream	Removal of both NO and NO <sub>2</sub> Requires 2 Columns. Lower CO <sub>2</sub> Recovery. NO in Vent Stream. Disposal of Liquid Streams Containing Nitrate Species.	NO Stripping – Proven NO <sub>2</sub> Removal – Developmental
NO <sub>x</sub> Scrubbing	Chemical scrubbing solution (e.g., peroxide, ozone, potassium permanganate) to remove NO <sub>2</sub> from CO <sub>2</sub>	Chemical cost, disposal of liquid waste stream	Developmental



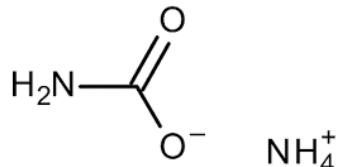
# NO<sub>x</sub> – Incidental Removal During CO<sub>2</sub> Purification

Process	Reported NO <sub>x</sub> Removal*	Key Challenges
NO <sub>x</sub> Removal in Compressor Condensate	Minimal at 10 psig (0.7 barg). ~ 60% at 216 psig (15 barg).	Limited by what dissolves in the water. NO <sub>2</sub> can be converted to NO and Nitric Acid in the Condensate and NO Can Desorb Back into the Gas.
NO <sub>x</sub> Removal in Dryer Beds	~ 15%.	NO <sub>x</sub> ends up in Regeneration Gas (Sometimes Recycled, Sometimes Vented)
NO <sub>x</sub> Removal in Compressor Oil	High Acid Number in Oil after 850 Hours of Operation.	Increased Oil Monitoring Requirements. Could Impact Oil Consumption and Compressor Reliability

\*Reference: Final Scientific Report, DOE award number: DE-FE0013163

# Other Contaminants Seen in CCUS (1 of 3)

- Corn Oil and Other Impurities in Ethanol  
Plant CO<sub>2</sub> Can Solidify with Changes in Pressure and Temperature.
- Compressor Oil in CO<sub>2</sub> and Produced Water (Condensate)
- Glycol Carry Over in CO<sub>2</sub> from TEG Dehydration Unit.
- Free Ammonia Gas can React with CO<sub>2</sub> to form ammonium carbamate salts.



Ammonium Carbamate<sub>(s)</sub>



Sulfur-containing Solid on liquid CO<sub>2</sub> pump impeller in Distillation Plant.

# Other Contaminants Seen in CCUS (2 of 3)

- ❑  $\text{SO}_2$  Oxidizes to  $\text{SO}_3$  and forms Liquid  $\text{H}_2\text{SO}_4$  Droplets (Aerosol) in DCC.
- ❑  $\text{SO}_3$  can act as Nucleation Site other Aerosol Formation.
- ❑ Aerosols are Very Difficult to Remove.
- ❑ Technology Options:
  - Alkaline Sorbent Injection Upstream of PM Control Removes up to 99% of  $\text{H}_2\text{SO}_4$ .
  - Wet ESP removes sulfuric acid and non-sulfuric acid aerosols (90% one field, 99% two fields).
  - Absorber Water Wash Design for Improved Aerosol Removal
  - Combinations of the above technologies.

# Other Contaminants Seen in CCUS (3 of 3)

- ❑ Mercury (Hg) found in Coal-Fired Flue Gas is a Hazardous Air Pollutant.
- ❑ Mercury is Very Detrimental to Brazed Aluminum Used in “Cold Box” Heat Exchangers.

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## Removal Options

- ❑ “Co-Benefits” Removal: SCR and FGD systems reduce Hg Concentrations.
- ❑ Activated Carbon Injection upstream of PM removal device.
- ❑ Halogen addition to the coal to oxidize mercury and remove in FGD.

# Team Acknowledgement

Contributors	Slide and Content Contributions
Joe Lundeen, P.E.	Dehydration
Brad Piggott, P.E.	$O_2$ , Liquefaction and Distillation, Catalyst Systems
Katherine Dombrowski, P.E.	$SO_2$ , $SO_3$ , $H_2SO_4$ , Mercury (Hg)
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